Fast direct methods for molecular electrostatics Kenneth L. Ho

Introduction

Electrostatic interactions are well-known to play an important role in determining protein structure and function, with charge interaction energies reaching the order of 10-100times the characteristic energy kT. The electrostatic potential describing the effect of such interactions is therefore an important quantity to consider in the analysis of proteins, whether arising through the study of natural biomolecules or through the design and engineering of synthetic ones.

The calculation of the electrostatic potential of a solvated biomolecule is computationally intensive, and its cost increases very quickly with the size of the protein under consideration. Thus, fast algorithms are necessary to make the electrostatic analysis of macro-Current state-of-the-art methods molecules feasible. are typically iterative, coupling fast matrix-vector multiplication routines with conjugate gradient-type schemes.

Here, we consider instead direct methods, which enjoy several significant advantages over their iterative counterparts, and describe our recent progress in the development of fast direct solvers for molecular electrostatics.

Direct solvers of $Ax = b$
(1) Reduced sensitivity to the con- ditioning of <i>A</i> .
(2) Fast application of A^{-1} , once

computed, to multiple vectors

3) Efficiency at handling low-rank perturbations of A.

This is joint work with Leslie Greengard and Zydrunas Gimbutas.

	Pote	ntial equations
$\begin{split} & \Sigma \\ & \Omega_1 \\ & \Omega_2 \\ & \nu \\ & \varphi \\ & \varepsilon = \begin{cases} \varepsilon_1 \text{ in } \Omega_1 \\ & \varepsilon_2 \text{ in } \Omega_2 \end{cases} \\ & q_i \\ & \boldsymbol{r}_i \\ & \lambda = \sqrt{\frac{2Ie^2}{\varepsilon_2 kT}} \end{split}$	 molecular surface protein interior ionic solution unit normal to Σ electrostatic potential dielectric permittivity charge strengths charge locations inverse Debye length 	The electrostatic potential is assumed to satisfy Poisson's equation in the protein inte- rior, with sources given by the fixed charges of the pro- tein; and the linearized Poisson- Boltzmann equation in the sur-
$K_L = \frac{1}{4\pi r}$	Laplace potential	describes the effect of ionic
$K_Y = \frac{e^{-\lambda r}}{4\pi r}$ $f * g$ σ μ	Yukawa potential convolution on Σ inner charge density outer charge density	screening in the solvent. Thus, we have the interface potentia problem

 $\nabla^{2}\varphi(\boldsymbol{r}) = \begin{cases} -(1/\varepsilon_{1})\sum_{i=1}^{n}q_{i}\delta(\boldsymbol{r}-\boldsymbol{r}_{i}), & \boldsymbol{r}\in\Omega_{1}\\ \lambda^{2}\varphi(\boldsymbol{r}), & \boldsymbol{r}\in\Omega_{2} \end{cases}$

with the jump conditions $[\varphi] = [\varepsilon(\partial \varphi / \partial \nu)] = 0$ on Σ . We use the single-layer potential solution representation

 $\varphi\left(\boldsymbol{r}\right) = \begin{cases} \left(K_{L} * \sigma\right)\left(\boldsymbol{r}\right) + \varphi_{0}\left(\boldsymbol{r}\right), & \boldsymbol{r} \in \Omega_{1} \\ \left(K_{Y} * \mu\right)\left(\boldsymbol{r}\right), & \boldsymbol{r} \in \Omega_{2} \end{cases}$

where

On discretizing using collocation, with the convolution integrals evaluated by quadrature over a triangle mesh of Σ , we obtain a linear system Ax = b, where x is a vector of the inner and outer surface charge densities at the collocation points.



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Figure 1: Potential equations and solution representaion.

$$\varphi_0(\mathbf{r}) = \sum_{i=1}^n q_i K_L(\mathbf{r} - \mathbf{r}_i)$$

is the potential due to the fixed charges, to obtain a boundary integral system in the unknown surface charge densities σ and μ . The jump conditions then become

Boundary integral systems 1) Dimensional reduction from volume to surface (2) High accuracy from faithful representation of geometry

$$K_L * \sigma - K_Y * \mu = \varphi_0,$$

$$\varepsilon_1 \left(\frac{\sigma}{2} + \frac{\partial K_L}{\partial \nu} * \sigma \right) - \varepsilon_2 \left(-\frac{\mu}{2} + \frac{\partial K_Y}{\partial \nu} * \mu \right) = -\varepsilon_1 \frac{\partial \varphi_0}{\partial \nu}.$$

Accelerated solution

We now deal with the linear algebraic problem of solving Ax = b. In general, the $N \times N$ system matrix A is dense. However, smoothness of the integral operators allows the high-accuracy representation of well-separated interactions by low-rank approximations. We exploit this by *compressing* A to obtain a $K \times K$ skeleton matrix, where typically $K \ll N$. Since direct inversion of an $m \times m$ matrix takes $\mathcal{O}(m^3)$ operations, this dramatically reduces the work required to compute A^{-1} .



Figure 2: One-level matrix compression.





This process can be applied recursively by compressing the skeleton after a suitable reordering of the variables by clustering. This leads to a multilevel matrix inversion scheme.

Once the surface charge densities have been determined, the electrostatic potential may be evaluated at any point in space by using the single-layer potential representations. This step may be accelerated as well by using the fast multipole method (FMM) to apply the discretized integral operator.

to scale as $\mathcal{O}(N \log^2 N)$.

Application to pK_a calculations

As an illustration of the power of direct methods, we outline the problem of computing the pK_a of a titratable residue in a protein. Briefly, the pK_a measures the strength of an acid in solution by assessing its protonation state. The pK_a of key residues, like Asp, Glu, Lys, Arg, and His, often give insight into the stability and reactivity of a protein.



Figure 4: Multilevel matrix compression.

The overall computational cost of the algorithm is expected

Table 1: Memory requirements.			
atrix	Dimensions	Memory use	
ll ompressed	$200,000 \times 200,000$ $2,000 \times 2,000$	320 Gb 32 Mb	

Figure 5: Protonation of aspartate.

$\mathbf{p}K_a \ \mathbf{p}K_i^{intr} \ heta$	acid dissociation constant shifted pK_a of <i>i</i> th site protonation state of protein
$ heta_i = egin{cases} 1 ext{ if HA} \\ 0 ext{ if A}^- \end{cases}$	protonation state of i th site
$\langle \cdot \rangle$	state average
p(heta)	probability of θ
W(heta)	electrostatic energy of θ
k	Boltzmann's constant
T	temeprature
Ζ	partition function

where

$$p(\theta) = \frac{1}{Z} \exp\left[\ln 10 \sum_{i} \theta_{i}\right]$$

is the probability of protonation state θ of the protein.

Since a protein typically has multiple titrating sites, the calculation of the titration curve $\langle \theta_i \rangle$ as a function of the pH, from which the pK_a may be derived, requires the sampling of many different protonation states θ . While an iterative solver must solve each case essentially independently, a direct solver can take advantage of the fact that θ only changes the vector b in the system Ax = b, so once A^{-1} has been computed, each solve $x = A^{-1}b$ may be obtained at minimal cost by applying A^{-1} , which requires only $\mathcal{O}(N)$ operations by using the FMM.

Furthermore, protein conformations are dependent on the electrostatic environment. Thus, different choices of θ are expected to induce local deformations of the surface geometry. This constitutes a low-rank perturbation of A, which a direct method can handle very efficiently, given A^{-1} , by forming the Schur complement.

Table 2: Fast application of matrix inverse.				Molecular surface triangulation wit
Folerance	Skeletons	Factorization	Application	N = 20,600. Times are for a single processor 1.9 GHz workstation. For
$\epsilon = 10^{-2}$	K = 7,049 K = 1.506	20 min	0.3 s	comparision, a FMM-based iterative a gorithm requires several seconds for
$\epsilon = 10^{-2}$	$\kappa = 1,390$	10 11111	0.1 S	each solve.

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$\epsilon = 10^{-2}$	K = 7,049	20 min	0.3 s	comparision, a FMM-based iterative a
$\epsilon = 10^{-1}$	K = 1,596	10 min	0.1 s	each solve.

Current progress

We currently have an implementation of the Poisson/linearized Poisson-Boltzmann solver in Fortran, as well as an interface to the matrix compression libraries. Multilevel matrix compression is under development, as is an interface to UMFPACK for fast matrix factorization.

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The pK_a can be calculated from the Henderson-Hasselbalch equation as the pH at which the acid is half-protonated on average. This involves taking a Boltzmann sum

 $\left\langle \theta_{i}\right\rangle =\sum\theta_{i}p\left(\theta\right),$

(p K_i^{intr} $-\mathbf{p}H$)



